

LYCOPERSICONOLIDE, A STEROID LACTONE FROM TOMATO ROOTS

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Key Word Index—*Lycopersicon* sp.; Solanaceae; roots; steroid lactone; lycopersiconolide; ^1H NMR; ^{13}C NMR.

Abstract—Lycopersiconolide has been characterized as (20S)-3 β ,16 β ,20-trihydroxy-5 α -pregnane-20-carboxylic acid 22,16-lactone.

INTRODUCTION

From roots of the tomato Taibyō Shinko No. 1 (*Lycopersicon esculentum* \times *L. hirsutum*, hybrid, Takii Co. Ltd.) a new steroid lactone, lycopersiconolide (1), was isolated as colourless needles.

RESULTS AND DISCUSSION

The ^{13}C NMR spectrum of 1 contained signals for three methyls, eight methylenes, seven methines and four quaternary carbons. The chemical shift values of the carbon atoms of rings A, B and C of compound 1 (Table 1) and tomatidine (4) [1] agreed very well, whereas those of ring D showed some differences. By comparison with the ^{13}C NMR values of several other steroids [2, 3] it was presumed that 1 possessed a 3 β -hydroxy-5 α -androstane ring system. The rest of the ^{13}C NMR signals [δ 74.9 (quaternary), 19.1 (Me), and 179.6 (quaternary)] and the IR spectrum indicated the presence of two hydroxy groups (3500, 3300 cm^{-1}) and a γ -lactone group (1780, 1750 cm^{-1}) and suggested that a α -hydroxy- α -methyl- γ -lactone moiety was bound to ring D. The molecular formula expected from these data was $\text{C}_{22}\text{H}_{34}\text{O}_4$ and in agreement with this the FDMS of 1 exhibited the molecular ion peak at m/z 362. The location of the lactone was evident from the ^1H NMR spectrum in which the signals at δ 1.52 (3H, s, —Me), 2.06 (1H, d, J = 5.9 Hz, $\text{HC}-$), and 5.09 (1H, ddd, J = 7.8, 6.2, 3.9 Hz, $\text{HC}-\text{O}$) were assigned to the protons of the γ -lactone moiety. These data support the structure of 3 β ,16,20-trihydroxy-5 α -pregnane-20-carboxylic acid 22,16-lactone for 1.

In order to determine the configuration of C-20 and of the D–E ring junction, NOE difference spectra were measured. A strong NOE was observed between 21-Me (1.52 ppm) and 18-Me (0.80 ppm). The irradiations, however, had no effect on H-16 and H-17. This indicated that the H-atoms at C-16 and C-17 were in the α -configuration, and that C-20 had S-configuration.

Acetylation of 1 with acetic anhydride/pyridine at room temperature afforded a monoacetate (2) and a diacetate (3). The spectral and physicochemical data of 2 and 3 completely agree with those of the mono and diacetate

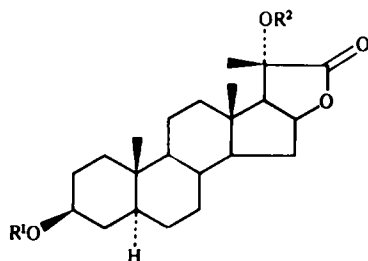
derived from (20S)-hydroxytigogenin acetate with concentrated HNO_3 [4]. Hence, 1 was confirmed to be (20S)-3 β ,16 β ,20-trihydroxy-5 α -pregnane-20-carboxylic acid 22,16-lactone.

The occurrence of three other lactones of this type has been reported in *Solanum* species; solanolide (5) from *S. hispidum* [5], vespertilin (6) [6] and 20S-hydroxyvespertilin (7) [4] from *S. vespertilio*.

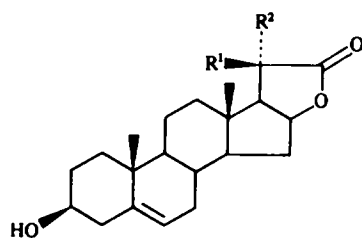
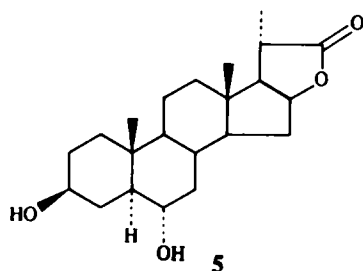
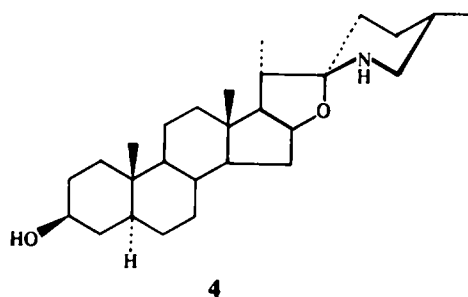
Table 1. ^{13}C NMR chemical shifts of lycopersiconolide (1) (67.8 MHz, $\text{CD}_3\text{OD}-\text{CDCl}_3$ (1:1), TMS as internal standard)

Carbon No.	Chemical shift, δ
1	37.6
2	31.4
3	71.2
4	38.1
5	45.4
6	29.1
7	32.2
8	35.2
9	54.7
10	36.1
11	20.8
12	39.3
13	41.1
14	56.4
15	32.5
16	83.5
17	64.2
18	14.0
19	12.6
20	74.9
21	19.1
22	179.6

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- 1 $R^1 = R^2 = H$
 2 $R^1 = Ac, R^2 = H$
 3 $R^1 = R^2 = Ac$



- 6 $R^1 = H, R^2 = Me$
 7 $R^1 = Me, R^2 = OH$

EXPERIMENTAL

Isolation. Fresh roots (7.5 kg) taken from Taibyo Sinko No. 1 tomato plants grown at the Agricultural Experimental Farm of Hokkaido University, were finely cut and extracted with 70% EtOH (55 l). The extract was filtered and concentrated to 0.5 l

under reduced pressure below 40°. The aq. residue was diluted with H₂O to 1 l and extracted with Et₂O (3 × 1 l). The Et₂O extracts were bulked, dried (Na₂SO₄), and the solvent removed *in vacuo*. The residue (10 g) was subjected to CC on silica gel and eluted with CHCl₃, CHCl₃-MeOH (19:1), (9:1), (4:1), (7:3) and MeOH to yield 13 fractions [monitoring with UV (280 nm)]. The 7th fraction eluted with CHCl₃-MeOH (19:1) was further chromatographed on a silica gel column with CHCl₃-MeOH (99:1). Compound 1 (50 mg) was obtained as colourless needles from CHCl₃-MeOH. Mp 295–297°; $[\alpha]_D^{20} - 48^\circ$ (MeOH; *c* 0.088); FDMS *m/z* (rel. int.): 362 [M]⁺ (5), 361 [M-H]⁺ (9), 360 [M-2H]⁺ (14), 346 (15), 318 (100); EIMS *m/z* (rel. int.): 360 [M-2H]⁺ (0.4), 342 [360-H₂O]⁺ (0.7), 329 (1), 318 [M-CO₂]⁺ (3), 303 (41), 273 (4), 257 (6), 138 (11), 107 (61), 84 (100); IR ν_{max}^{KBr} cm⁻¹: 3500, 3300, 1780, 1750; ¹H NMR [500 MHz, CD₃OD-CDCl₃ (1:1), TMS]; δ 0.80 (3H, s, H₃-18), 0.84 (3H, s, H₃-19), 1.52 (3H, s, H₃-21), 2.06 (1H, d, *J* = 5.9 Hz, H-17), 2.23 (1H, m, H-15 α), 3.54 (1H, m, H-3), 5.09 (1H, ddd, *J* = 7.8, 6.2, 3.9 Hz, H-16).

Monoacetate (2) and diacetate (3) of 1. A soln of 1 (18 mg) in Ac₂O (0.5 ml) and C₅H₅N (0.5 ml) was kept at room temp. for 12 hr. CC of the residue on silica gel developed with CHCl₃ gave a monoacetate (2, 15 mg) and a diacetate (3, 4 mg). **Monoacetate (2)**, mp 280–282°, lit. 282–284° [4], needles (CHCl₃-*n*-Hexane); $[\alpha]_D^{20} - 58^\circ$ (CHCl₃; *c* 0.24), lit. -54° (CHCl₃) [4]; FDMS *m/z* (rel. int.): 405 [MH]⁺ (15), 360 (35), 344 (100); EIMS *m/z* (rel. int.): 404 [M]⁺ (0.07), 386 [M-H₂O]⁺ (0.9), 360 (2), 345 (28), 285 (6), 257 (6), 217 (11), 107 (29), 84 (100), 43 (94); IR ν_{max}^{KBr} cm⁻¹: 3450, 1780 (shoulder), 1760, 1740, 1240; ¹H NMR (270 MHz, CDCl₃, TMS): δ 0.80 (3H, s, H₃-18), 0.83 (3H, s, H₃-19), 1.57 (3H, s, H₃-21), 2.02 (3H, s, H₃-Ac), 2.08 (1H, d, *J* = 6.2 Hz, H-17), 2.23 (1H, m, H-15 α), 4.68 (1H, m, H-3), 5.07 (1H, ddd, *J* = 7.8, 6.2, 3.9 Hz, H-16). **Diacetate (3)**, mp 270–272°, lit. 261–262° [4], needles (CHCl₃-*n*-Hexane); $[\alpha]_D^{20} - 27^\circ$ (CHCl₃; *c* 0.11), lit. -26° (CHCl₃) [4]; FDMS *m/z* (rel. int.): 446 [M]⁺ (22), 386 (100); EIMS *m/z* (rel. int.): 446 [M]⁺ (0.1), 404 (0.5), 386 (23), 360 (22), 345 (65), 311 (8), 285 (10), 257 (13), 216 (19), 107 (31), 84 (32), 43 (100); IR ν_{max}^{KBr} cm⁻¹: 1790, 1740, 1730, 1250, 1220; ¹H NMR (270 MHz, CDCl₃): δ 0.80 (3H, s, H₃-18), 0.83 (3H, s, H₃-19), 1.79 (3H, s, H₃-21), 2.02 (3H, s, H₃-Ac), 2.04 (3H, s, H₃-Ac), 2.23 (1H, m, H-15 α), 2.70 (1H, d, *J* = 6.2 Hz, H-17), 4.68 (1H, m, H-3), 4.95 (1H, ddd, *J* = 7.8, 6.2, 3.9 Hz, H-16).

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REFERENCES

1. Radeaglia, R., Adam, G. and Ripperger, H. (1977) *Tetrahedron Letters* 903.
2. Blunt, J. W. and Stothers, J. B. (1977) *Org. Magn. Reson.* 9, 439.
3. Agrawal, P. K., Jain, D. C., Gupta, R. K. and Thakur, R. S. (1985) *Phytochemistry* 24, 2479.
4. González, A. G., Freire, R., Francisco, C. G., Salazar, J. A. and Suárez, E. (1973) *Tetrahedron* 29, 1731.
5. Chakravarty, A. K., Das, B. and Pakrashi, S. C. (1982) *Phytochemistry* 21, 2083.
6. González, A. G., Francisco, C. G., Barreira, R. F. and Lopez, E. S. (1971) *An. Quim.* 67, 433; *Chem. Abst.* 75, 115890n (1971).